

Name: \_\_\_\_\_

## PHYS 1220-02 Quiz 1

You may use an 8.5"×11" note sheet written on both sides and a calculator. You have 50 minutes.

Please write your answers in the boxes provided. Show your work outside the boxes.

1. Xylenes are high-boiling hydrocarbon solvents. They are normally sold as a mixture of *o*-xylene, *m*-xylene, and *p*-xylene. *O*-Xylene's boiling temperature is 144°C, *m*-xylene's boiling temperature is 139°C, and *p*-xylene's boiling temperature is 138°C.

- a. (1 pt) What is *o*-xylene's boiling temperature in kelvin?

- b. (1 pt) What is the difference between the boiling temperatures of *o*-xylene and *p*-xylene in kelvin?

2. The viaduct (highway bridge) on Third Street just south of Laramie over the train tracks is 170.0 meters long in the summertime at a temperature of 25°C. The viaduct is made of steel, which has a thermal coefficient of length expansion of  $\alpha = 11 \times 10^{-6}/^\circ\text{C}$ .

- a. (1 pt) In the winter, when the span is at a temperature of  $-20^\circ\text{C}$ , is it shorter or longer than it is in the summer? (Select the best answer by filling the circle).

☐ Shorter

☐ Longer

- b. (2 pt) In the winter, at a temperature of  $-20^\circ\text{C}$ , how much (length) shorter or longer is the span compared to its length at 25°C?

3. (1 pt) The engineering plastic PEEK has a thermal coefficient of length expansion of  $\alpha = 55 \times 10^{-6}/^\circ\text{C}$ . What is its coefficient of volume expansion?

4. (3 pt) Xylenes, with a thermal coefficient of volume expansion of  $\beta = 10 \times 10^{-4}/^{\circ}\text{C}$ , is transported in a glass bottle,  $\beta = 27 \times 10^{-6}/^{\circ}\text{C}$ . 4.00 liters of xylenes are packed in a glass bottle with an interior volume of 4.10 liters at  $20^{\circ}\text{C}$ . The bottle of xylenes is shipped in a hot truck, reaching a temperature of  $45^{\circ}\text{C}$ . What volume of the bottle is not occupied by xylenes at  $45^{\circ}\text{C}$ ? If the volume of xylenes exceeds the capacity of the bottle at  $45^{\circ}\text{C}$ , give a negative number.

5. (1 pt) “Heat rises” applies to which heat transfer mechanism? Select all correct answers by filling their squares.

☐ Conduction

☐ Convection

☐ Radiation

6. (1 pt) Which heat transfer mechanism operates in a vacuum? Select all correct answers by filling their squares.

☐ Conduction

☐ Convection

☐ Radiation

7. Very roughly, the Antarctic ice sheet has a mass of  $2.6 \times 10^{19}$  kilograms and a temperature of  $-15^{\circ}\text{C}$ . The latent heat of melting ice is  $334 \text{ kJ/kg}$ , and the specific heat capacity of ice is  $2.10 \text{ kJ/(kg}\cdot^{\circ}\text{C)}$ . The energy input that Earth receives from the Sun is around  $1.66 \times 10^{17}$  watts.

- a. (1 pt) What heat input would raise the temperature of the Antarctic ice sheet to  $0^{\circ}\text{C}$ ?

- b. (1 pt) What further heat input would melt all the ice in the Antarctic ice sheet?

8. (1 pt) An aluminum pot sits on a hot stove burner. The bottom of the pot is 0.500 cm thick, and is circular with an area of  $0.0154 \text{ m}^2$ . The burner is maintained at a temperature of  $98.0^\circ\text{C}$ , and the pot contains water boiling at  $95.0^\circ\text{C}$ . The thermal conductivity of aluminum is  $250 \text{ W}/(\text{m}\cdot\text{K})$ . What is the heat current (rate of heat transfer) through the bottom of the

pot?

9. (1 pt) A bag of potato chips is packed and sealed containing 1.50 liters of air at sea level pressure of  $1.01 \times 10^5 \text{ Pa}$  and temperature of  $27^\circ\text{C}$ . It is shipped to Laramie, where the temperature is  $21^\circ\text{C}$  and the pressure is  $7.5 \times 10^4 \text{ Pa}$ . At this temperature and pressure, what is the volume of the air in the bag?

10. (1 pt) A sealed gallon jug containing air inside a heated house at a pressure of  $1.01 \times 10^5 \text{ Pa}$  and temperature of  $20.0^\circ\text{C}$  is taken outdoors in winter, so that the air in the jug cools to  $-20.0^\circ\text{C}$ . What is the pressure of the air in the jug at this temperature? You may neglect any change in the volume of the jug.